Design of "model-friendly" turbulent non-premixed jet burners for C $_{2+}\,$ hydrocarbon fuels

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TABLE III. Flow parameters of the ßames used in our experiments.

$ \begin{array}{cccc} Q^a & L_{f,e}{}^c & L_{f,m}{}^d & FR_{f}{}^e & FR_{p}{}^f & U {}^g & T_{f}{}^h & T_{p}{}^i \\ Fuel & Re {}(kW) & Fr_{f}{}^b {}(mm) {}(mm) {}(slpm) {}(slpm) {}(m/s) {} $										
C2H4	10 000) 12.0	1.0	652	2 730	13.2	4.45	0.60	21	21
	15000	18.0	1.4	755	775	19.8	6.68			
	20 000	24.0	1.9	832	830	26.4	8.91			
	25 000	30.0	2.4	890	861	33.0	11.14			
JP-8	20 00	0 16.5	1.0	860	872	0.029	5 6.09	0.67	300	205

^aHeating value of the main fuel jet.

^bFroude number of ßame; for JP-8, assumes 40% radiation loss.

 $^{\rm c}\textsc{Estimated}$ visible ßame length; for JP-8, assumes 40% radiation loss.

^dMeasured visible ßame length.

eVolumetric ßow rate of main fuel jet (liquid ßow for JP-8).

^f Volumetric ßow rate of pilot ßame premixture.

gVelocity of coßow air.

^hTemperature of unburnt fuel.

ⁱTemperature of pilot ßame premixture.

2% of the heat release rate of the main fuel jet. Detailed ßow parameters are listed in Table. Note that these ßames with Fr_f ranging between 1.0 and 2.4 fall in the transitional regime, where both buoyancy and jet momentum are importants-ible ßame heights estimated from E(d.) generally match measured values well, conbrming the validity of using this relation.

III. PRELIMINARY CHARACTERIZATION

learned from these preliminary measurements, such as the ßame size, the frequency of occurrence of local extinction, and soot behavior.

A. Visual observation

In general, a sooty jet ßame has a blue soot-free region near the nozzle (Fig1(c)) followed by a soot-laden region downstream. Soot produced in the ßame is either consumed or escapes from the ßame as smoke. The length of the sootfree region depends on the fuel, the fuel jet diameter, and the fuel jet velocity. Figure7 presents fast-shutter photos of the four ethylene jet ßames. With the 1/1600 s exposure time, the soot-free region is nearly invisible and the sooty regions appear orange, from the broad-band soot radiation. An increase in jet velocity pushes the sooty edge (the bottom edge of the soot-containing region) downstream. The photographs of the ßames also show a progression in ßame wrinkling with increasing jeRe The ßame with a Reynolds number of 10000 shows smooth, large-scale vortex rollup, particularly as the distance from the nozzle increases (and buoyancy effects become relatively stronger). For ßames with higher the luminous ßame front shows more intense wrinkling and smaller characteristic ßame structures, reßecting a more turbulent flow Þeld. The flame also broadens radially with the increase irRe reßecting the increased rate of mixing between the fuel jet and the air coßow.

Since the JP-8 ßame uses a burner with a smaller fuel The open ßame conÞguration permits convenient accessube diameter than the ethylene ßames, there is limited value for both optical and probe-based measurements. The Þviæ comparing these ßames. Instead, we can compare the JP-8 canonical ßames were characterized with qualitative meth.ßame to an ethylene ßame established on the same (liquidods, including direct photography, OH LIF planar imaging fuel) burner with the same fuel jæe(20 000). As shown in and soot LII planar imaging. In addition, measurementsFig. 8, these two ßames look somewhat different. While the of the burner lip temperature boundary with line Rayleigh ethylene jet ßame has a fairly long blue, soot-free region near imaging are presented for ethylene ßames. Much can behe nozzle, the JP-8 ßame is almost completely devoid of

that, with the increase in Re the OH layers at this height change from smooth sheets (F&(a)) to rugged and kinked

FIG. 10. (Color online) Instantaneous false-color OH PLIF images of three turbulent ethylene jet ßaltes (to 000; b,Re= 20000; c,Re= 25000) and one JP-8 ßame (Re= 20000). The same intensity map was used to process all of the OH PLIF images.

local Rayleigh scattering cross section of the chemical species requent local extinction followed by reignition downstream. mix in this region. Flames remain attached to the burner lip until the fueRjet

A similar measurement of the temperature boundary wass increased beyond 35 000, when extinction becomes so senot performed for the JP-8 ßame, but the nearly identical devere that the ßame is partially attached or lifted off from the sign of the pilot plate for this burner as for the gas burner isburner. Clearly, this relationship between the ßame Reynolds expected to give equivalent temperature probles at the burnerumber and extinction frequency is consistent with our preexit plane. diction based on U_J/D as shown in Fig2. It is also noted

C. OH LIF imaging

structures (Figs10(b)and10(c)), indicating the enhanced interactions between turbulence and the ßame zone. As OH is an important ßame marker, planar imaging Figure 10(d) shows an OH ßuorescence image of the of OH is an effective method of capturing extinction events, JP-8 ßame over a similar range of near-burner heights as for which appear as breaks within otherwise continuous OH layers. To perform these measurements, a UV laser beam at 283.55 nm was generated from the frequency doubled output of a YAG-pumped dye laser. The beam was shaped into a laser

sheet with sheet-forming optics and subsequently intersected the ßame on an axial plane, exciting OH radicals. The OH ßuorescence was detected by an intensibed CCD camera (ICCD-MAX, Princeton Instruments) operating with a 100 ns exposure time using a 105-mm focal length, f/4.5 UV Nikkor lens and a band-pass Plter set that transmits 304D320 nm. This OH detection scheme provides good discrimination against ßame emission, laser scattering and ßuorescence from polycyclic aromatics, but accepts ßuorescence from single and doublering aromatics⁵⁹

Figure 10 shows false-color OH LIF images centered about 24 jet diameters downstream, where strong strain coupled with diminishing inßuence of the pilot ßame results in the highest probability of local ßame extinction. For ethylene ßames with a Reynolds number of 15000 or less, breakup of OH layers rarely occurs (Fig.0(a)). Breaks in the OH layers are occasionally apparent in the ethylene ßame will eaf 20 000 (Fig.10(b)), and become a frequent feature when the ßameRereaches 25000 and above (Filg0(c)), suggesting